



Development of new heavy and efficient scintillators for medical imaging and radiation detection

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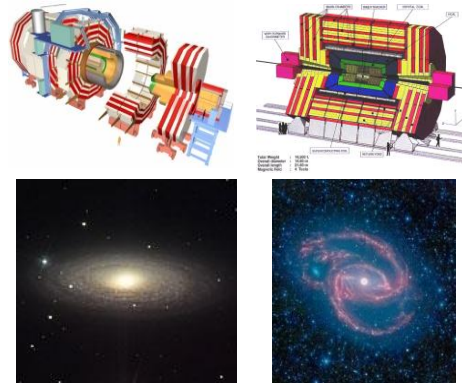
Application of Scintillators



Medical application



High energy physics



Security check



Astro-particle physics,

X-ray scanning

Nondestructive analysis

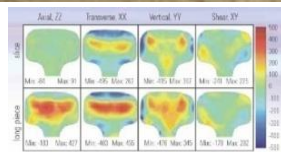


FIGURE 2
Stress maps (MPa) for the two investigated samples.

Board inspection



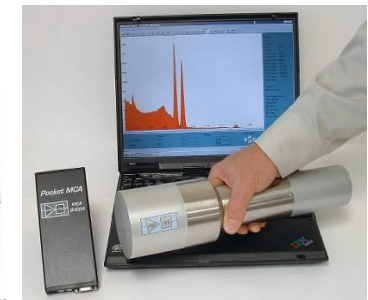
Radiation monitoring

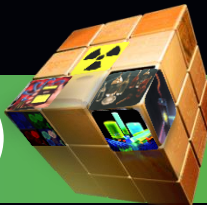
Scintillation Detectors



Scintillation Surveymeter

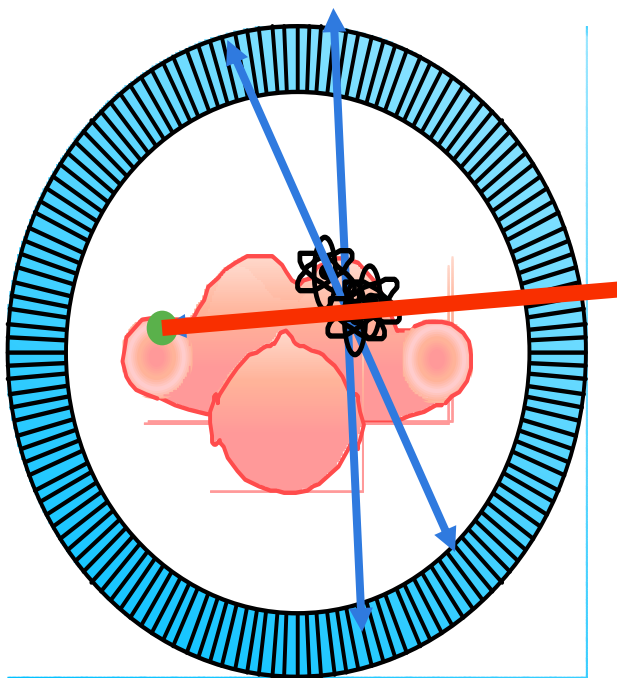
Nal(Tl) Spectrometer



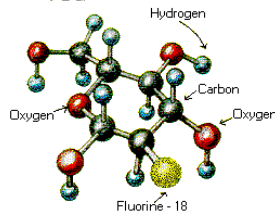


Positron Emission Tomography (PET)

Ring of Photon Detectors

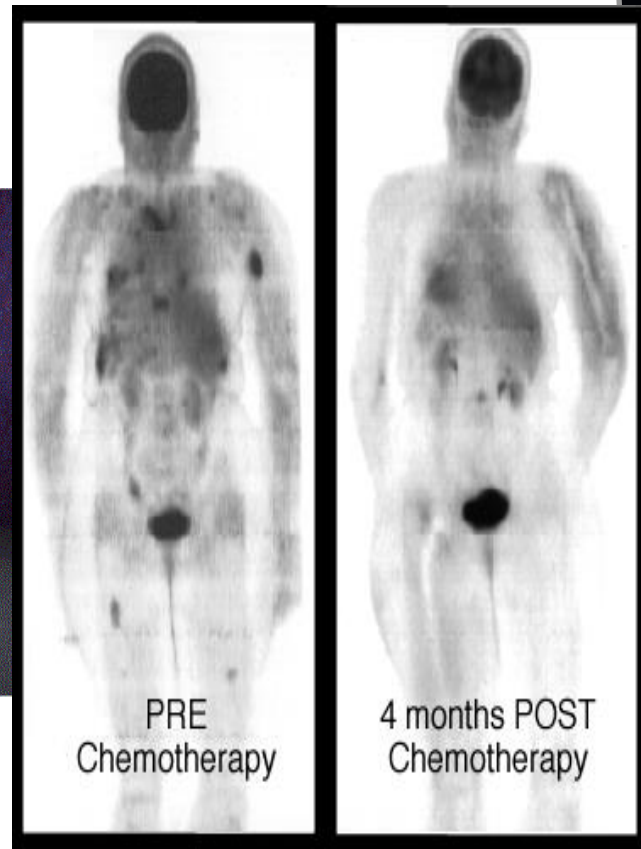


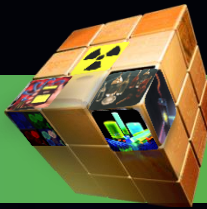
2-fluoro-
2-deoxy-D-glucose
"FDG"



POSITRACE

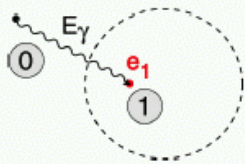
Dual Mode PET/CT
Oncology System





Detection Efficiency of X-ray and γ

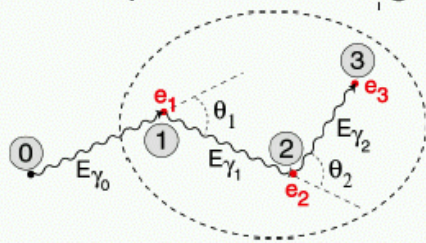
Photoelectric



Isolated hits

Probability of interaction depth

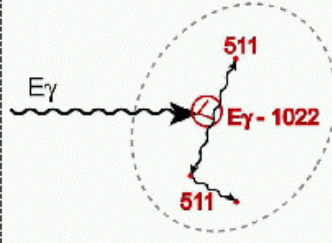
Compton Scattering



Angle/Energy

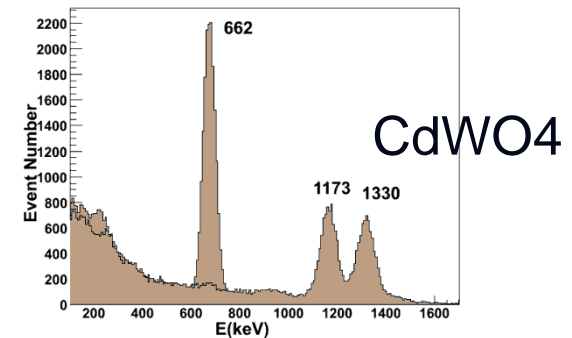
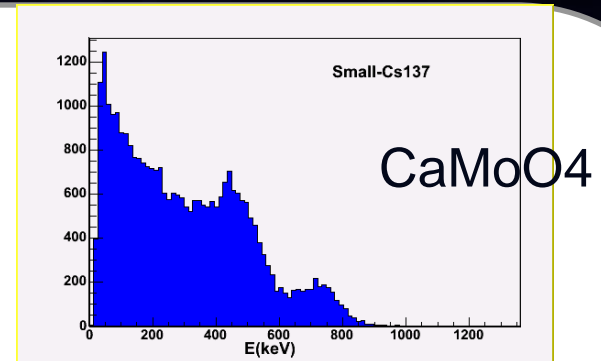
$$E_{\gamma'} = \frac{E_{\gamma}}{1 + \frac{E_{\gamma}}{m_0 c^2} (1 - \cos\theta)}$$

Pair Production



Pattern of Hits

$$E_{1st} = E_{\gamma} - 2mc^2$$



Photoelectric effect : $\rho Z_{\text{eff}}^{3-4}$

Compton scattering : ρ

Pair production : ρZ

High ρ & Z_{eff} !

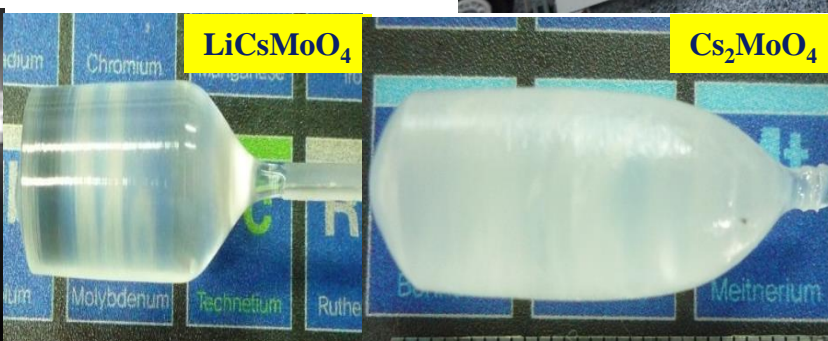
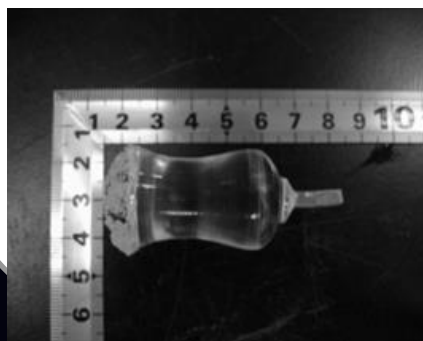
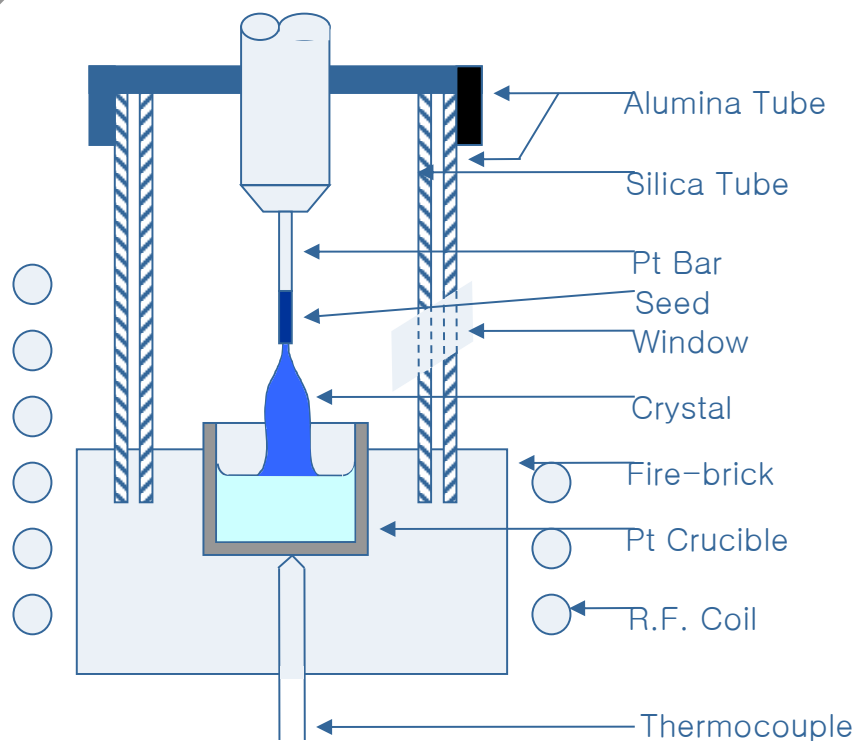
Motivation

- Our research work is to explore and develop new Tl-based inorganic halide scintillators for different application such as high energy and nuclear physics, radiation monitoring, homeland security as well as medical imaging.
- Alteration of the energy band gap, enhancement of Z_{eff} and density using Tl ion with high density ($\rho=11.8 \text{ g/cm}^3$) and high Z-number ($Z=81$). Disadvantage of Tl is highly toxic that extreme care should be taken. (ex: TlBr, KRS-5, KRS-6 CsI:Tl, NaI:Tl)
- Not only with intrinsic luminescence of Tl, with Ce^{3+} and Eu^{2+} doping for the improvement of scintillation properties.
- They could be applied PET, SPECT, Gamma Camera & CT

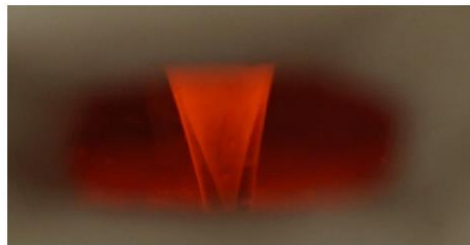


Crystal growing system with Czochalski method

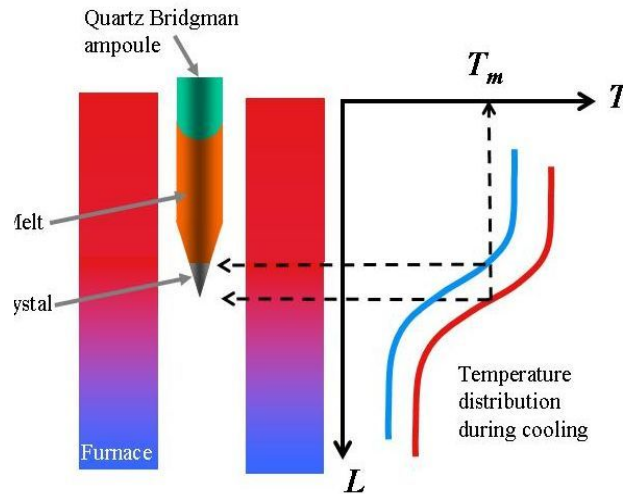
At KNU



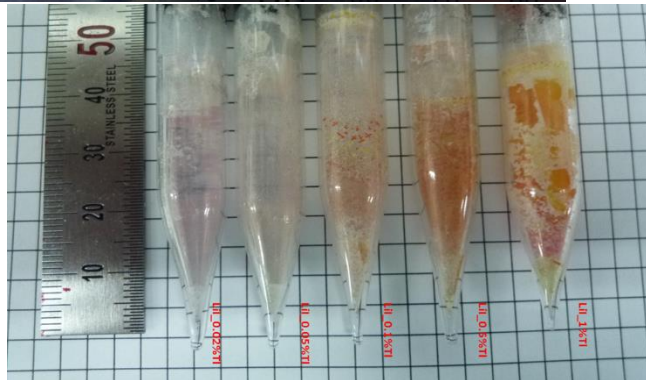
Bridgman Crystal Growing Methods



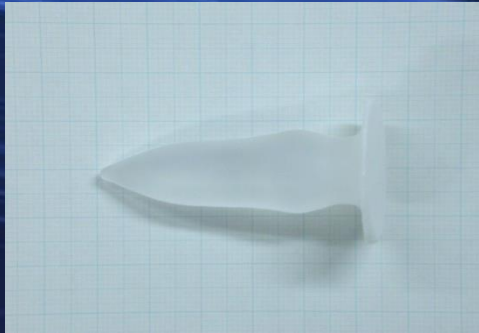
Crystal is growing in Bridgman furnace



At KNU



Crystals growing at KNU for 20 years



$\text{SrCl}_2(\text{Eu})$



CeCl_3



$\text{LaCl}_3(\text{Ce})$



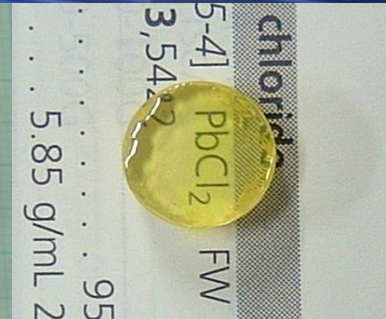
CeBr_3



CsI



$\text{Ba}_{0.2}\text{Sr}_{0.8}\text{Cl}_2$



$\text{PbCl}_2:\text{Eu}^{2+}$

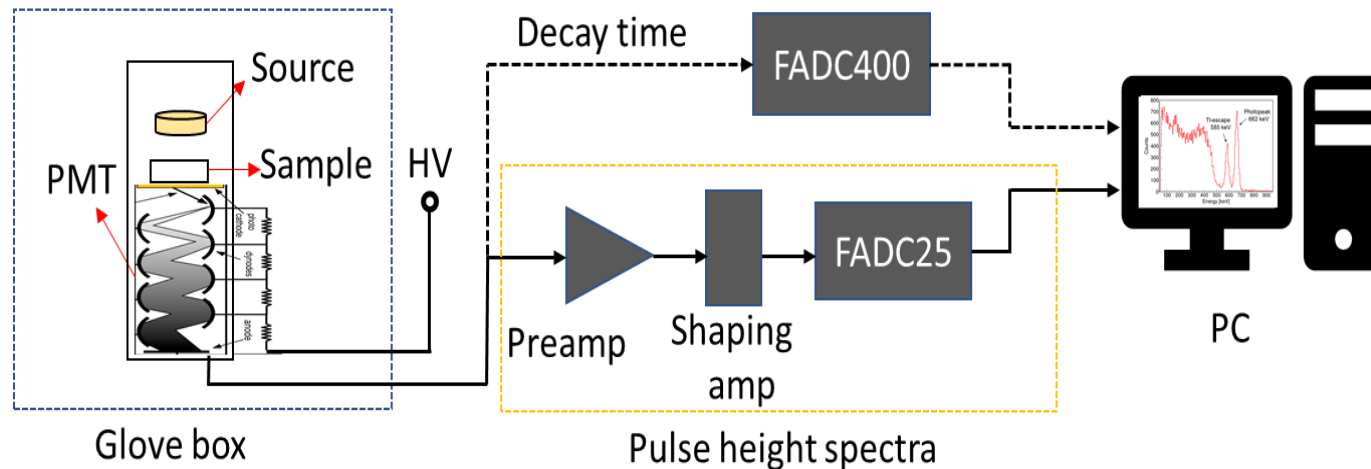


BGSO

- ◆ $\text{NaI}:\text{Ti}$, $\text{CsI}(\text{Ti}, \text{Co}^{3+}, \text{Na})$, BGO, BSO, BGSO, SrWO_4 , CaMoO_4 , SrMoO_4 et al.
- ◆ New material : BaSrCl_2 , CsCe_2Cl_7 , $\text{Cs}(\text{Rb})_2\text{Li}(\text{Na})\text{CeCl}_6$, $\text{Cs}_2\text{LiGd}(\text{Lu})\text{Cl}(\text{Br})_6:\text{Ce}$, $\text{Li}_6\text{Lu}(\text{Gd}, \text{Y})(\text{Bo}_3)_3$, $\text{NaGd}(\text{Wo}_4)_3$, **Tl-based scintillators** et al.



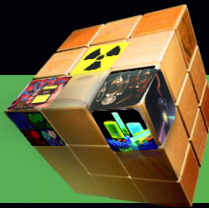
Experimental Setup



➤ For the γ -ray spectroscopy



Development of Elpasolite scintillators



Elpasolite scintillators : X_2YReZ_6 w/wo Ce^{3+} doping
($X=Cs, Rb$; $Y=Li, Na$; $Re=$ Rare Earth, and $Z=Cl, Br, I$)

Most studied examples: $Cs_2LiYCl_6:Ce$ (CLYC), $Cs_2LiLaCl_6:Ce$,
 $Cs_2LiLaBr_6:Ce$, $Cs_2LiLuCl_6:Ce$, $Rb_2LiYBr_6:Ce$ (neutron detection)

Studied by our group:

$Cs_2LiCeCl_6$, $Cs_2LiCeBr_6$, $Cs_2NaCeCl_6$, $Cs_2NaCeBr_6$
 $Cs_2LiGdCl_6:Ce$, $Cs_2LiGdBr_6:Ce$, $Cs_2NaGdCl_6:Ce$, $Cs_2NaGdBr_6:Ce$
 $Rb_2LiGdCl_6:Ce$, $Rb_2LiGdBr_6:Ce$, $Rb_2LiCeCl_6$, $Rb_2LiCeBr_6$
 $Cs_2LiLuBr_6:Ce \Rightarrow$ More than 13 publications and 4 patents

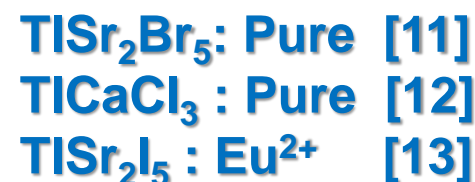
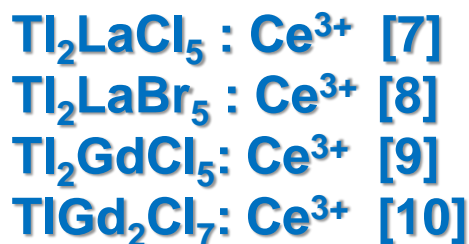
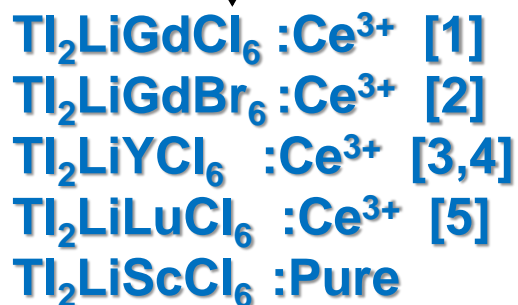
New High Z-number elpasolite scintillator?

X- \rightarrow TI ($Tl_2LiGdCl_6:Ce$, $Tl_2LiYCl_6:Ce$, $Tl_2LiGdBr_6$ & so on)

Discovery of Tl based scintillators



We started a pioneer work in 2009 on the development of Tl-based compounds and published $\text{Tl}_2\text{LiGdCl}_6:\text{Ce}^{3+}$ paper as a first Tl-based scintillator in 2015 and presented $\text{Tl}_2\text{LiYCl}_6:\text{Ce}^{3+}$ SCINT2015.



[11] G. Rooh et al., Opt. Mater., 73 (2017) 523.

[12] A. Khan et al., Rad. Measurement., 107 (2017) 115.

[13] H.J. Kim et al., Opt. Mater., 82 (2018) 8

[1] H.J. Kim et al., J. Lumin., 164 (2015) 86–89.

[2] H. J. Kim et al., Rad. Measurement., 90 (2016) 279–281.

[3] H. J. Kim et al., IEEE Trans. Nucl. Sci., 63 (2) (2016) 439.

[4] G. Rooh et al., J. Cryst. Growth, 459 (2017) 163–166.

[5] G. Rooh et al., J. Lumin., 187 (2017) 347–351.

[7] H.J. Kim et al., J. Lumin., 186 (2017) 219–222.

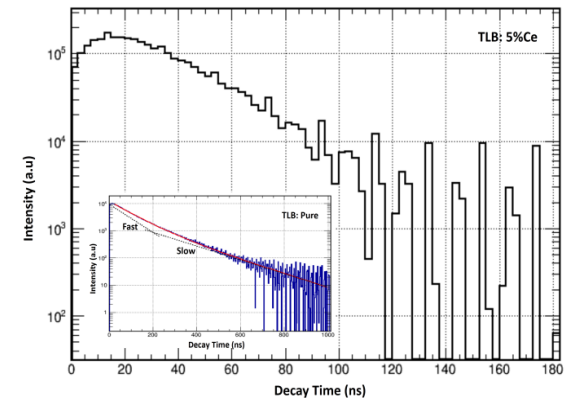
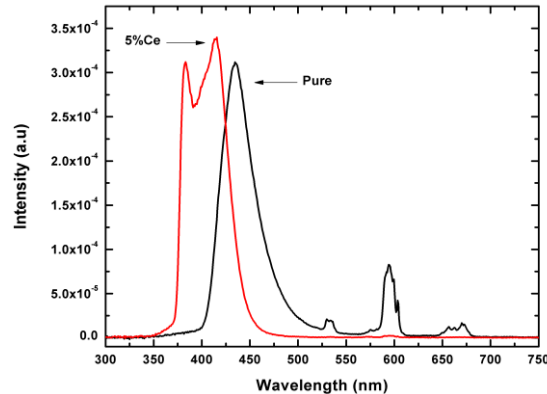
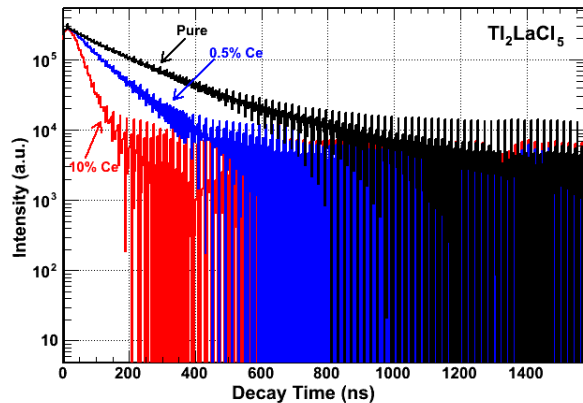
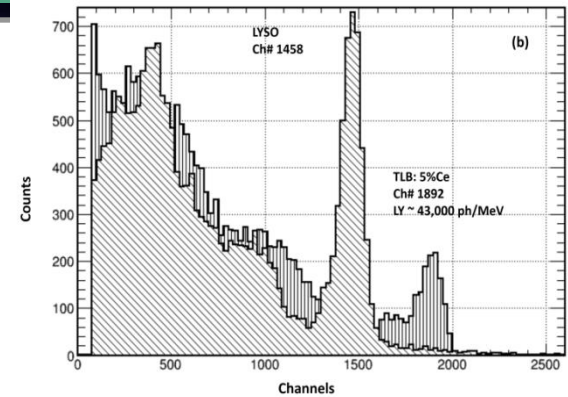
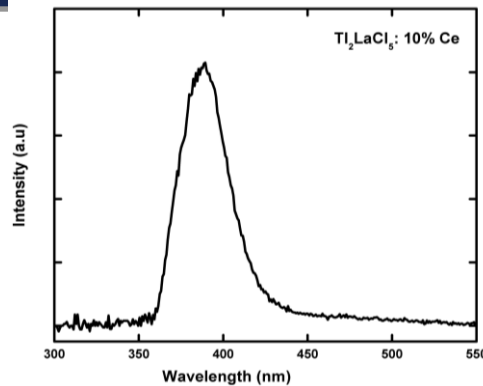
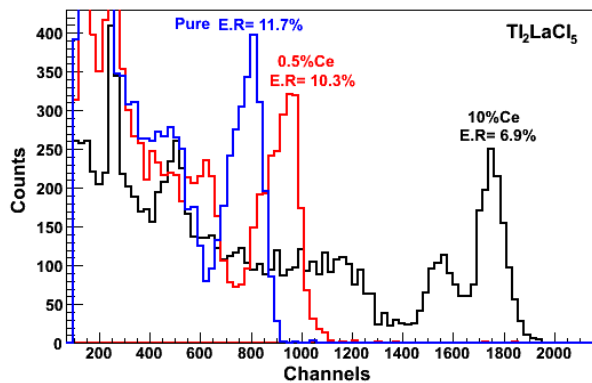
[8] H.J. Kim et al., NIMA., 849 (2017) 72–75.

[9] G. Rooh et al., IEEE TNS., 65 (8) (2018) 2157

[10] A. Khan et al., IEEE TNS., 65 (8) (2018) 2152

2 US and 1 EU patents, 1 is under review

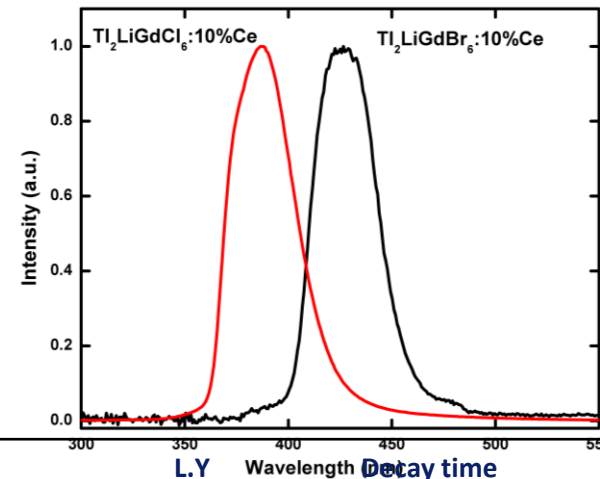
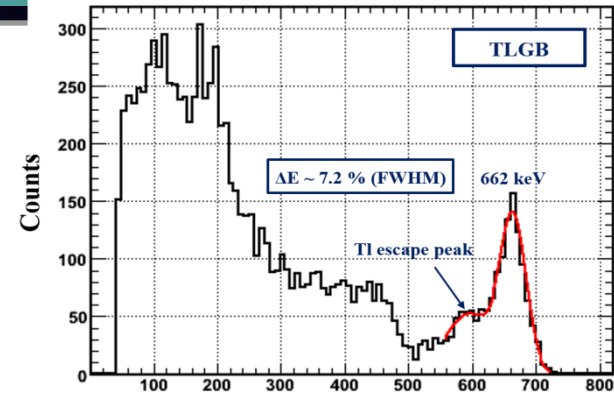
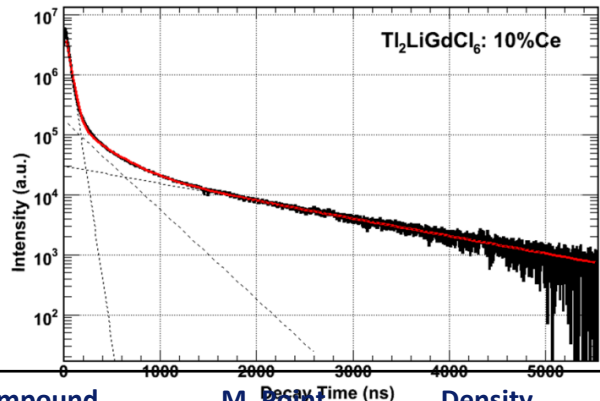
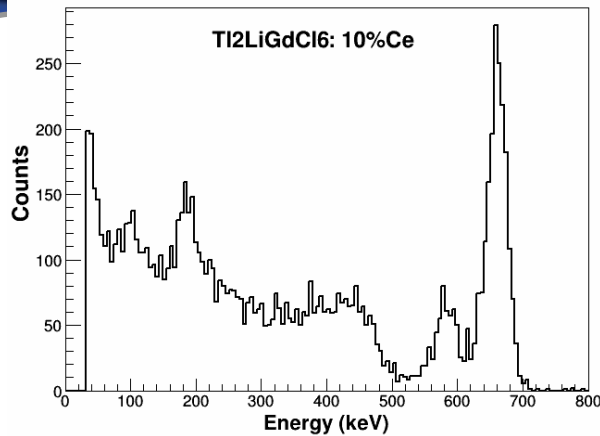
Ce-doped Ti_2LaCl_5 and Ti_2LaBr_5 Scintillators



Compound	M.P (°C)	Density g/cm ³	Z _{eff}	E.R	LY (ph/MeV)	Decay time (ns)	Ref.
Ti_2LaCl_5 : 10% Ce	510	5.20	71	6.9%	51,000	31(87%) + 111	[6]
Ti_2LaBr_5 : 5% Ce	548	5.90	67	6.3%	43,000	25	[7]
LaBr_3 : Ce^{3+}	783	5.08	46.9	3.0%	61,000	32	[10]



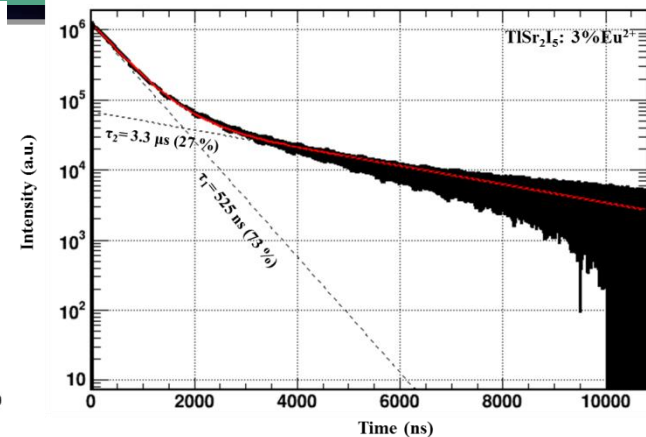
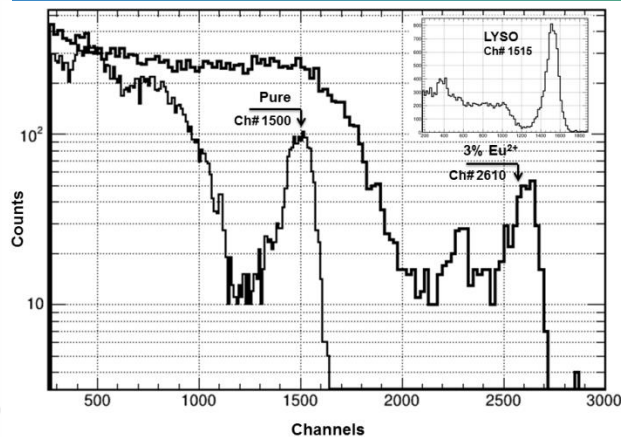
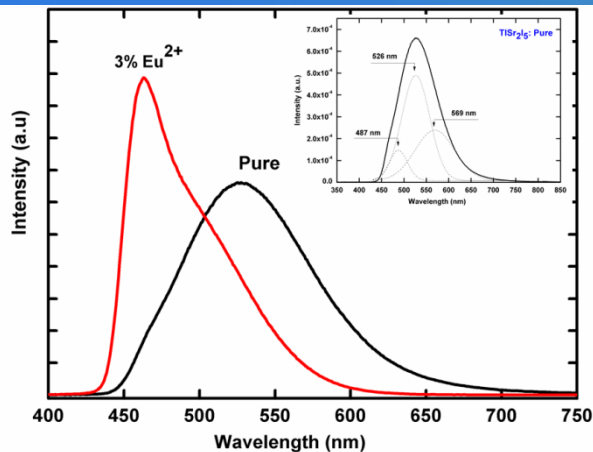
Ce-doped $\text{Tl}_2\text{LiGdCl}_6$ single crystal



Compound	M. Point (°C)	Density g/cm ³	Z _{eff}	E.R @662 keV	L.Y (ph/MeV)	Decay time (ns)	Ref.
$\text{Tl}_2\text{LiGdCl}_6$	540	4.75	70	4.6%	58,000	34 (81%)	[1]
$\text{Tl}_2\text{LiGdBr}_6$: Pure	450	5.30	66	7.2%	27,000	56+slow	
LaBr_3 : Ce^{3+}	783	5.08	46.9	3.0%	61,000	32	[11]

[11] P. Dorenbos et al., IEEE Trans. Nucl. Sci., NS-51 (2004) 1289.

Eu²⁺ doped TlSr₂I₅

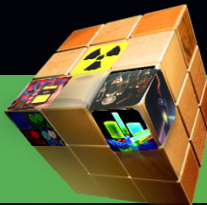


Compound	M.P (°C)	Density g/cm ³	Z _{eff}	E.R (FWHM)	L.Y (ph/MeV)	Decay time (ns)
TlSr ₂ I ₅ : 3%Eu ²⁺	480	5.30	61	4.2%	70,000	525 + slow
SrI ₂ : Eu ²⁺	538	4.6	49	3%	120,000	1200+slow [14]
KSr ₂ I ₅ : Eu ²⁺	471	4.39	50	2.4%	94,000	990 + slow [15]

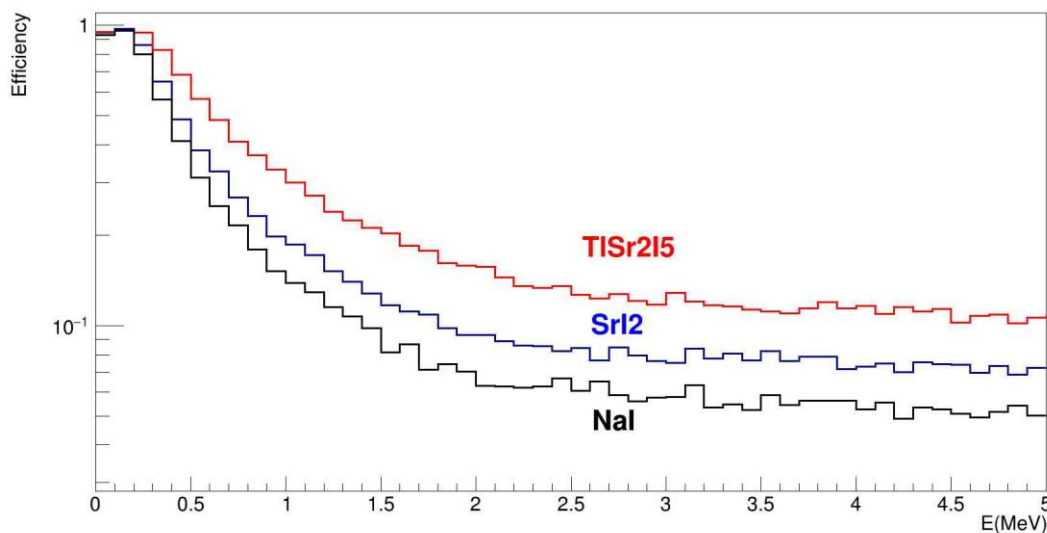
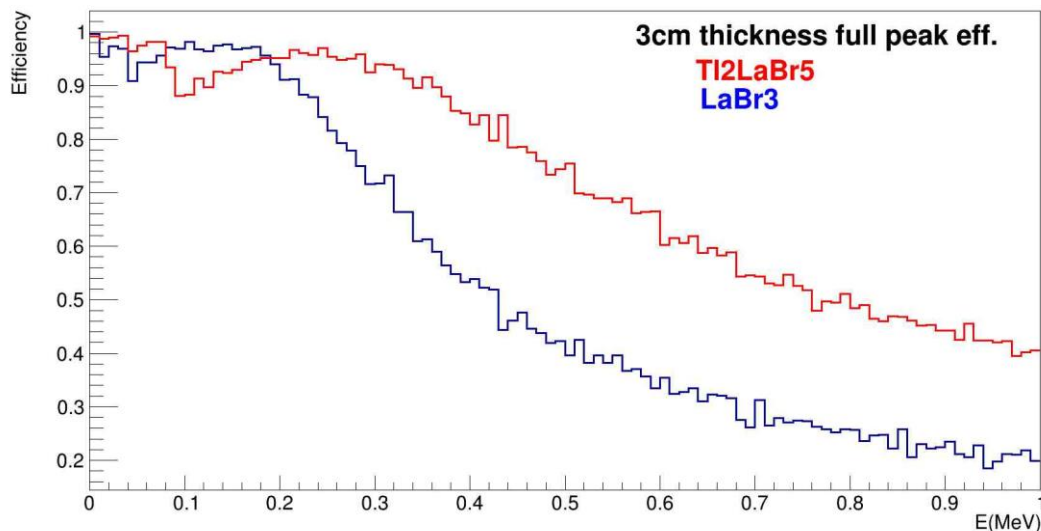
[14] Y. T. Wuet al., Cryst. Growth Des., 15 (2015) 3929.

[15] L. Stand et al., Nucl. Instrum. Methods Phys. Res. A 780 (2015) 40.

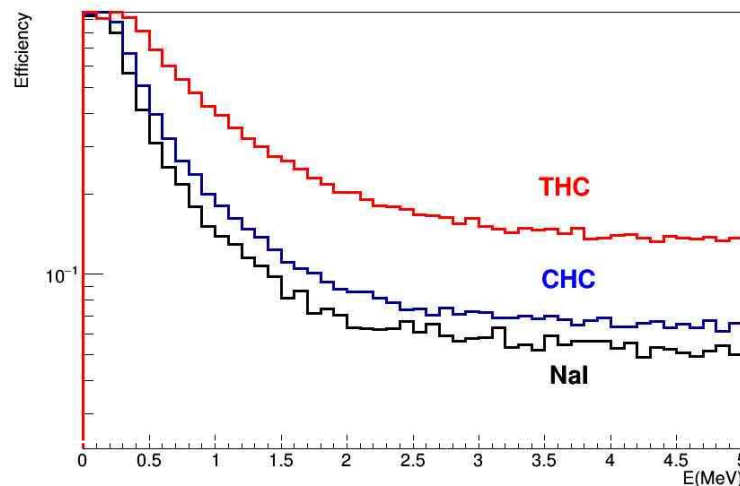
- Improvement is expected with material purification, optimized with Eu²⁺-doping concentration and with co-doping.



Full peak efficiency comparison

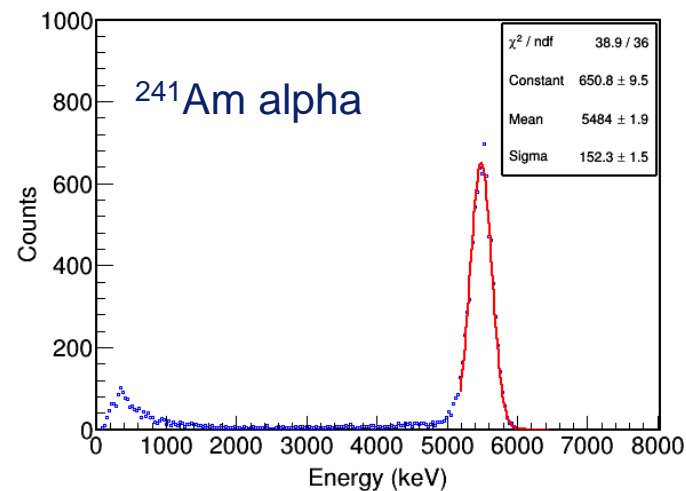
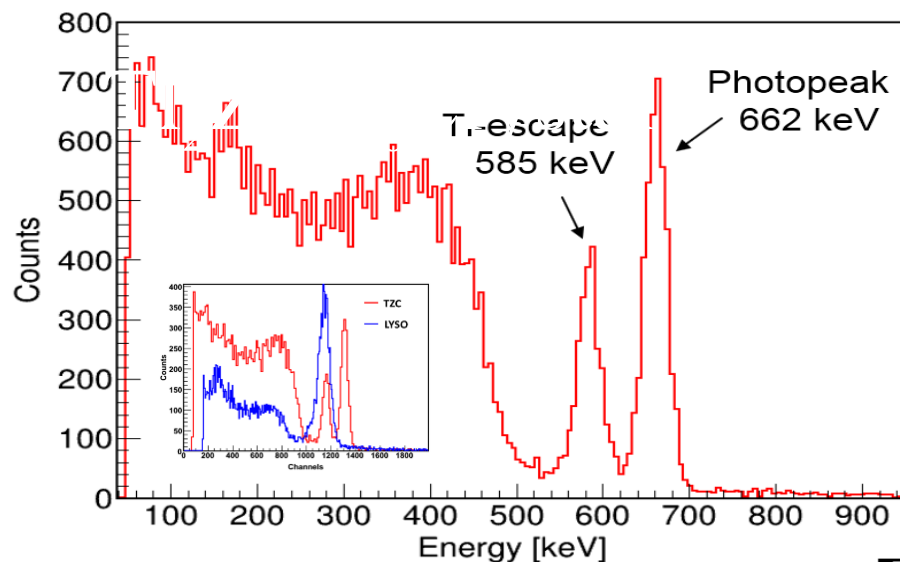


New crystal scintillators 2018



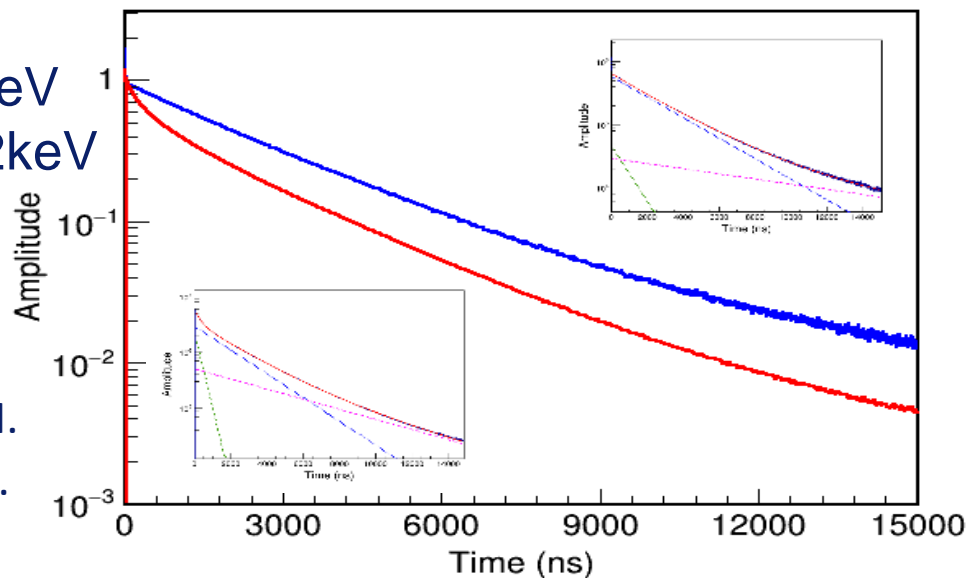
Scintillation material	Effective atomic number	Density (g/cm ³)	Light yield (ph/MeV)	Major decay time (ns)
NaI(Tl)	50.8	3.67	38000	250
CsI(Tl)	54	4.51	52000	1000
BGO	75.2	7.13	9000	300
LYSO	65	7.1	33000	40
Cs ₂ HfCl ₆ ⁽¹⁾	58	3.86	54000	4100
Tl₂ZrCl₆⁽²⁾	69	4.65	47000	2700
Tl₂HfCl₆	71	5.25	32000	1000

Tl₂ZrCl₆ crystal



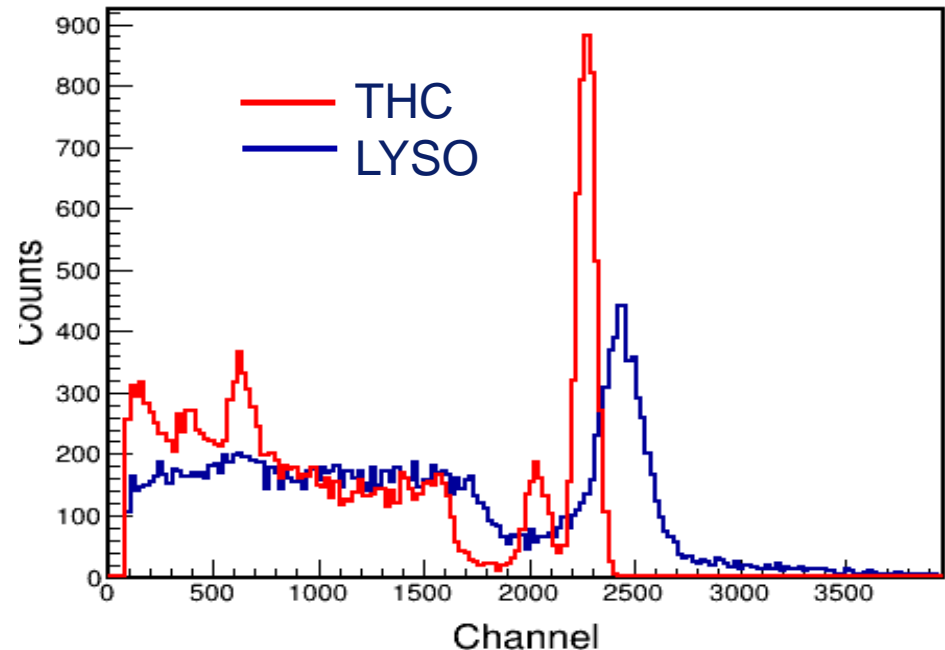
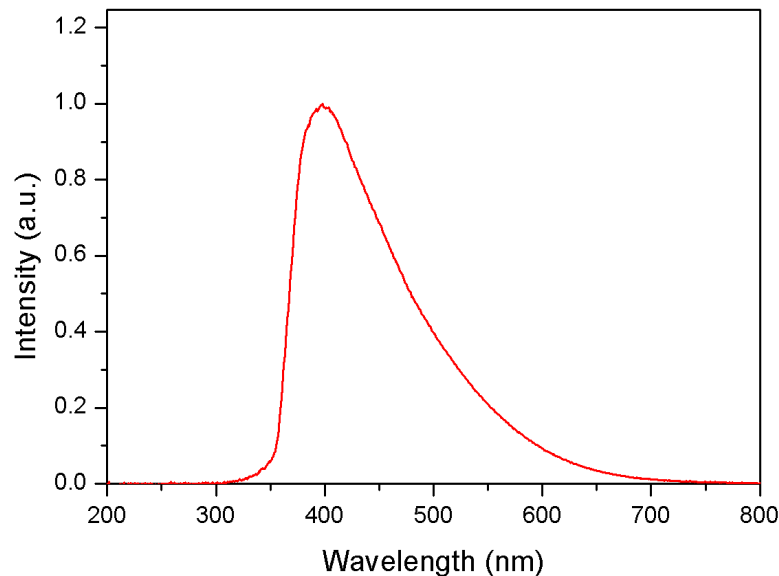
Light yield : 47.000 photons/MeV
 Energy resolution : 4.3% @ 662keV
 Main decay time :

² Q.V. Phan, H.J. Kim, G. Rooh, and S.H. Kim, J. Alloys Compd. **766**, 326 (2018).



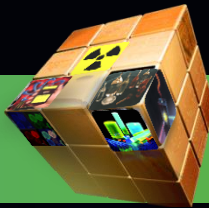


Ti_2HfCl_6 pulse height spectra



- Emission peak at 400 nm
- Light yield : 32.000 photons/MeV
- Energy resolution : 4.0% @ 662keV
- Better than Cs_2HfCl_6

Summary and future perspect



- New Tl-based scintillators are discovered and reported.
- Preliminary results showed promising performance with high light yield, high effective Z-number, fast decay time, good energy resolution and moderate density.
- Excellent detection efficiency of X- and gamma rays due to the high effective Z-number can be used for the radiation detection (ex: $\text{TlSr}_2\text{I}_5:\text{Eu}$) and medical imaging (ex: $\text{Tl}_2\text{LaCl}_5(\text{Br}_5):\text{Ce}$) while Li-based scintillator can be used for neutron detection (ex: $\text{Tl}_2\text{LiYCl}_6:\text{Ce}$).
- Since, optimization of the growth condition, dopant concentration, co-doping and purification are under way, therefore, further improvement of the scintillation performance is expected.

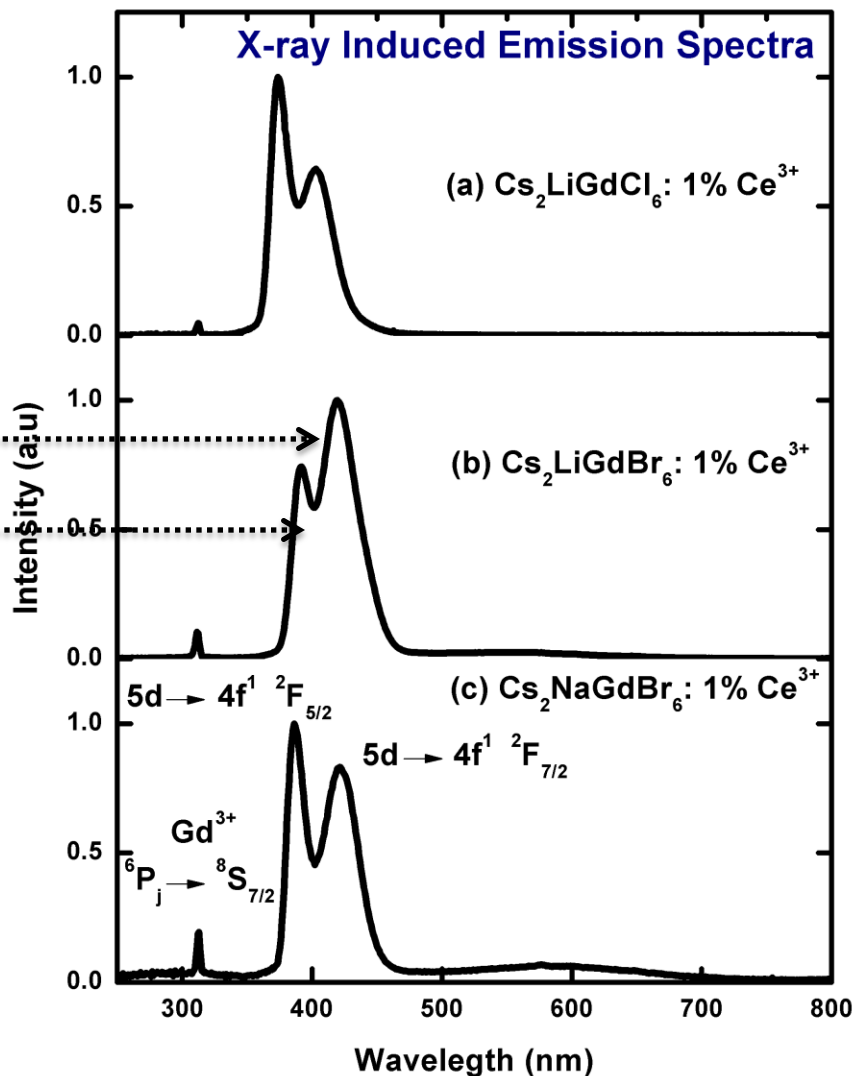
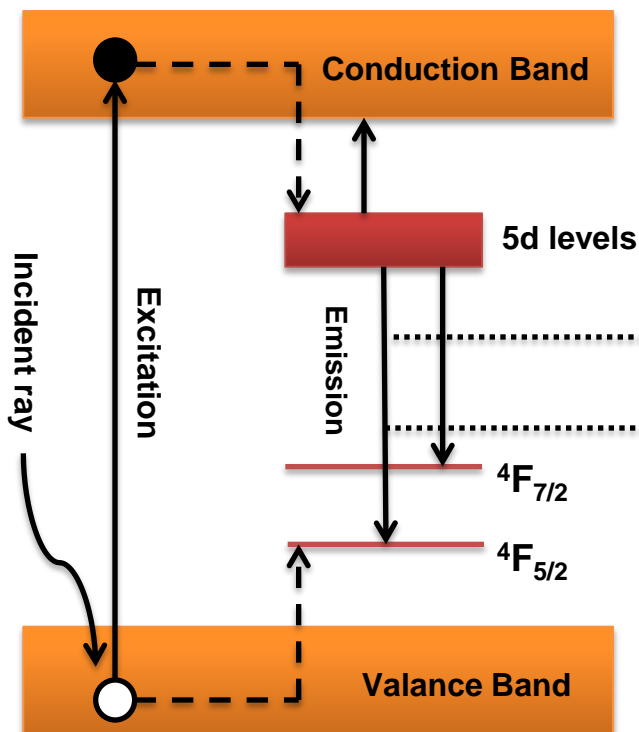


Thank you for your
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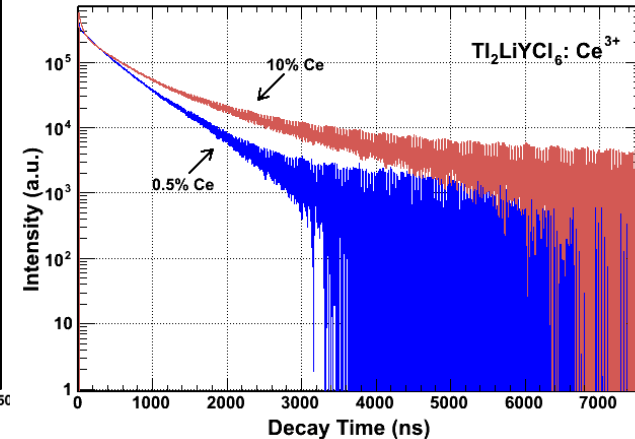
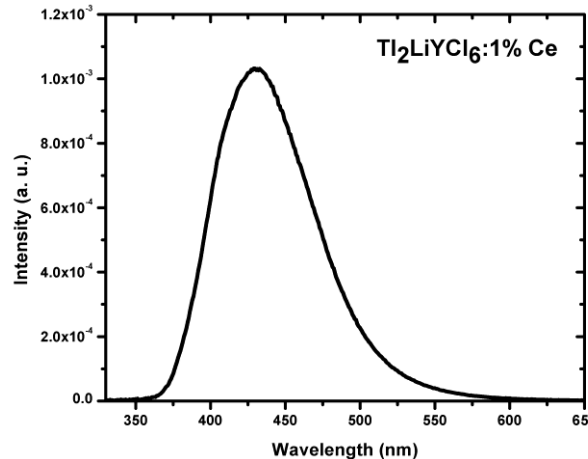
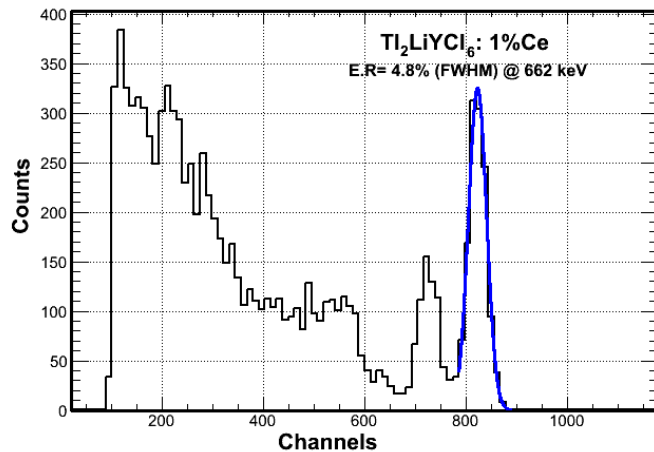
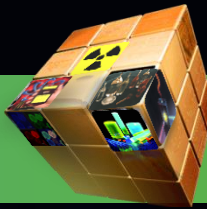


X-ray Induced Emission Spectra

Ce³⁺ Emission



Ce-doped $\text{Ti}_2\text{LiYCl}_6$ (TLYC)



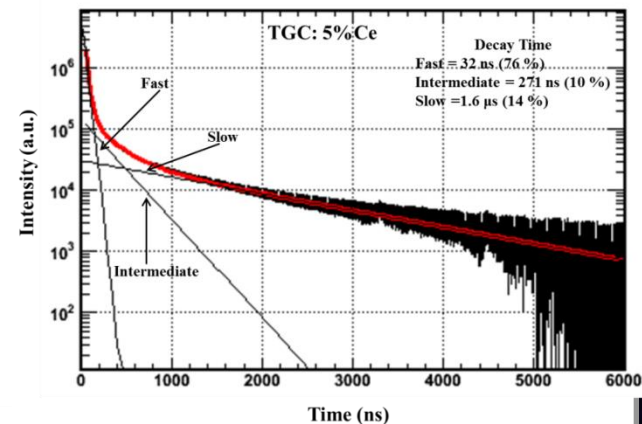
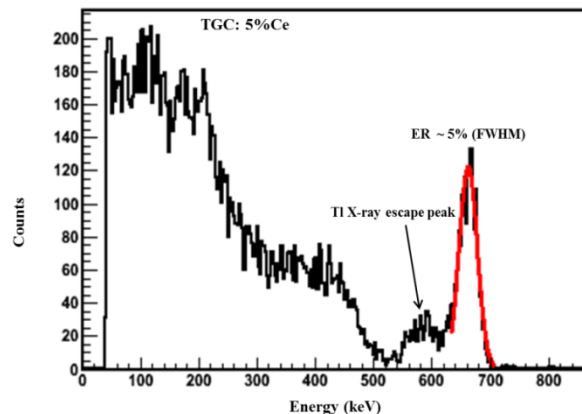
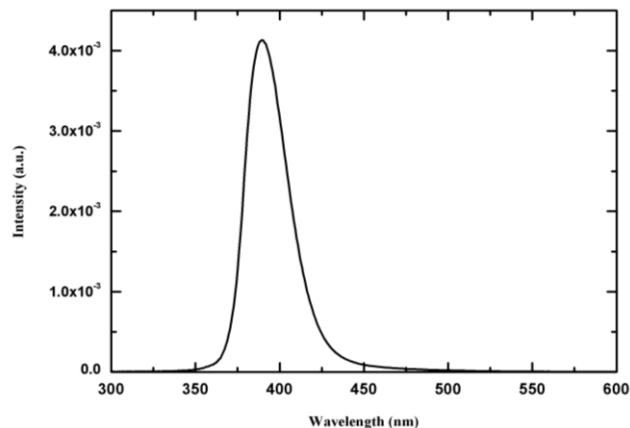
- [12] J. Glodo et al., IEEE Trans. Nucl. Sci., 55(3) (2008) 1206.
 [13] R. Hawrami et al., IEEE Trans. Nucl. Sci., 63(6) (2016) 2838.

Compound	M. Point (°C)	Density g/cm ³	Z _{eff}	E.R @662 keV	L.Y (ph/MeV)	Decay time (ns)	Ref.
$\text{Cs}_2\text{LiYCl}_6$	640	3.31	45	5.1%	20,000	129 + Slow	[12]
$\text{Ti}_2\text{LiYCl}_6$	490	4.58	69	4.8%	30,500	57(9%) + Slow	[3,4]

- TLYC of both gamma and thermal neutron detection efficiency is better than the $\text{Cs}_2\text{LiYCl}_6$ reported by RMD group [13].

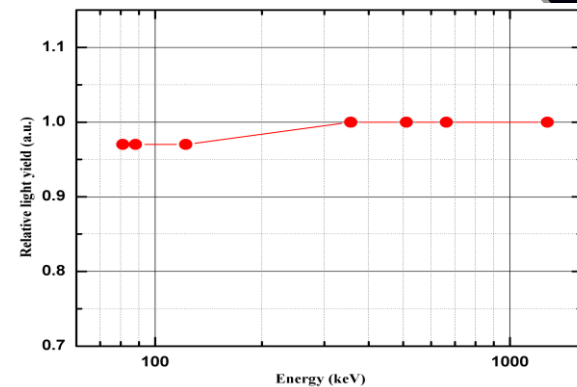
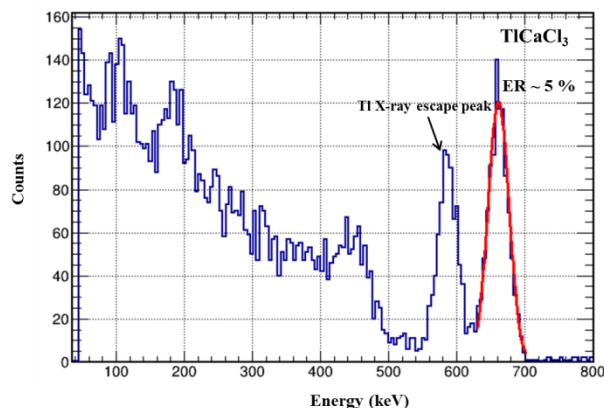
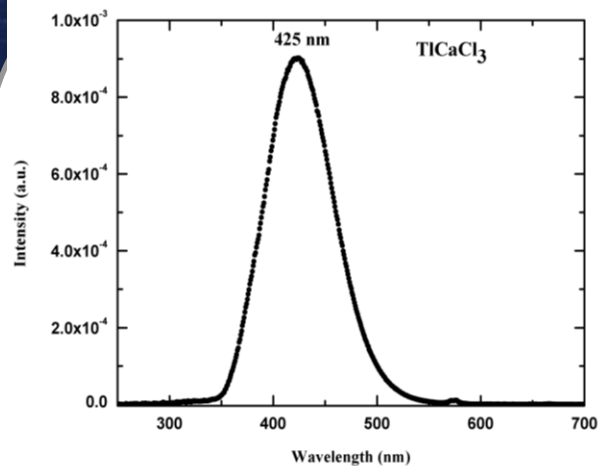


Ce-doped Tl_2GdCl_5

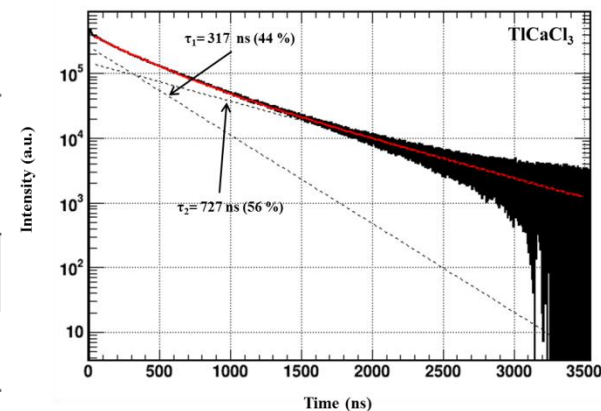


Compound	M.P (°C)	Density g/cm ³	Z _{eff}	E.R @662 keV	L.Y (ph/MeV)	Decay time (ns)
Tl_2GdCl_5	490	5.10	71	5.0%	53,600	32 (76%)

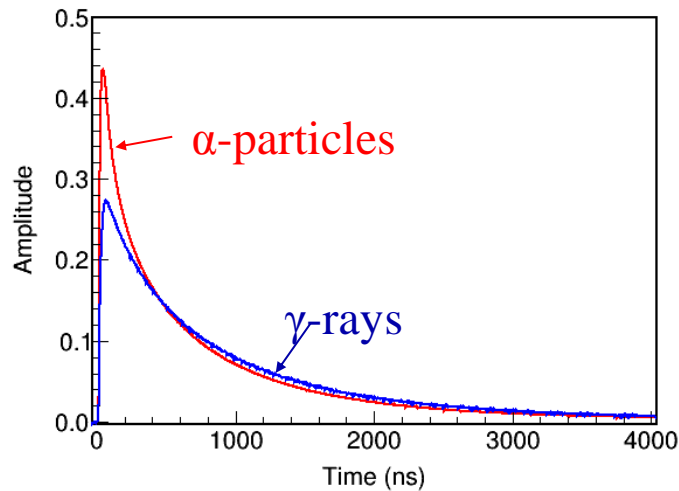
Perovskite type : TlCaCl_3 : Pure



Compound	M.P (°C)	Density g/cm ³	Z _{eff}	E.R (FWH M)	L.Y (ph/MeV)	Decay time (ns)
TlCaCl_3	680	3.80	65.5	5%	31,000	317 + 727
NaI: Tl	661	3.67	50.8	~ 7%	38,000	250



Tl₂HfCl₆ crystal



Type of irradiation	Decay constants (s) and their relative intensities (% of total)			
	$\tau_1 (I_1)$	$\tau_2 (I_2)$	$\tau_3 (I_3)$	$\tau_4 (I_4)$
γ	0.36(17.5)	1.04(61.2)	14.9(19.8)	
α	0.09(8.5)	0.46(28.6)	1.04(44.8)	11.2(17.9)

FOM=8.5 with Charge comparison method

